

The background of the slide is a close-up, high-contrast image of blue flames. The flames are bright blue and white at the tips, with darker blue and greyish-blue in the shadows, creating a dynamic and somewhat abstract pattern.

Dangerous Chemicals

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Some classes of Dangerous Chemicals

Pyrophorics

Corrosives



Acute toxins

Explosives

Carcinogens



Pyrophorics

a substance that will ignite spontaneously in air or upon exposure to water or water vapor

Organolithiums: *t*-BuLi, *n*-BuLi, MeLi

Organozincs: Et₂Zn

Organomagnesiums: EtMgBr

Organoaluminums: Me₃Al

Metal powders: Zn, Al, Mg, Ti, Ni, Pd

Alkali metals: K, Na

Metal hydrides: LiAlH₄

Non-metal alkyls: BR₃, PR₃, AsR₃



Often used in conjunction with flammables!

Working Safely with Pyrophorics

1.) Don't work alone!

- notify others, remind them of safety procedures
- beware of the location of the class D fire extinguisher

2.) Wear the right protection

- nitrile gloves or (nitrile + neoprene) gloves
- goggles or face shield
- buttoned-up fire-resistant lab coat

3.) Check the state of the reagent and your work space

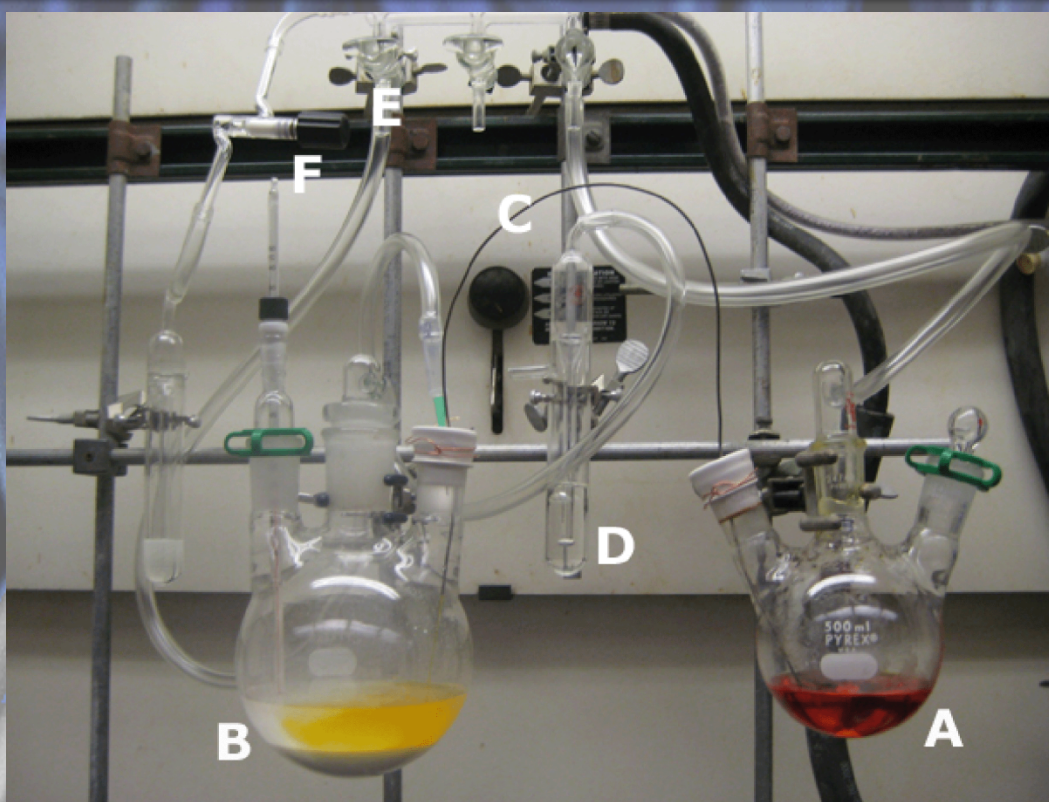
- corroded septum?
- excess flammable material or clutter in hood?

4.) Use the proper technique

- syringe or cannula?

Syringe and Cannulation Techniques

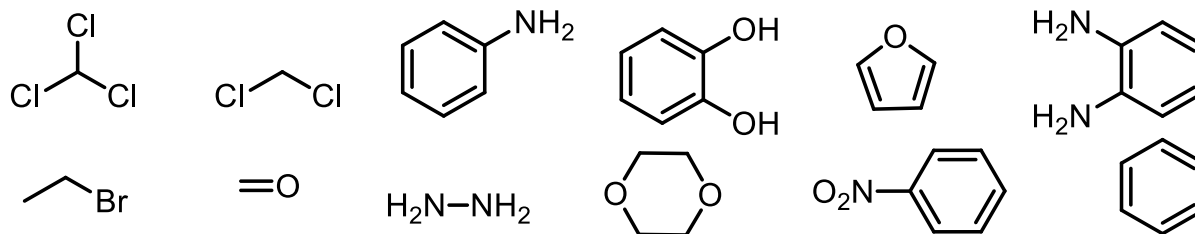
Watch UCLA pyrophoric safety video:
<http://www.youtube.com/watch?v=RaMXwNBAbxc>



Carcinogens

- 1.) Label known carcinogens as such.
- 2.) Store in carcinogen cabinet or in secondary containment.
- 3.) Change gloves after using carcinogens.

Some common carcinogens:



Explosives: Azides

What NOT to do with NaN_3 :

- use with chlorinated solvents (di- and tri- azidomethanes)
- mix with CS_2 , bromine, Bronsted acids, or heavy metals.

In general, “safe” azides conform to the formula:

$$(\#O + \#C)/(\#N) > \text{or} = 3$$

What to avoid with organic azides:

- concentrating to dryness
- putting under vacuum
- ground-glass joints
- mechanical shock
- distillation
- exposure to light and silica

What to definitely do:

- put in separate waste container
- avoid mixing with acid

Peroxides and peroxide formers

What to avoid :

- metals
- heat, friction, shock
- vacuum

How to hydrolize:

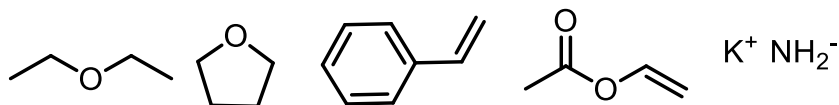
- Use 80 parts water, 20 parts NaOH, and small amount of surfactant
- Slowly pour peroxide in solution with stirring (use ~10X more alkaline solution)

How to test for peroxides:

Use starch-iodide paper.

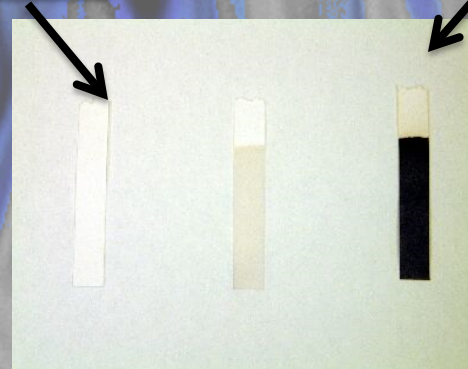
- immerse in solution for 1 sec
- breathe on strip for 30 sec

Common peroxide formers:



No
peroxide

High
peroxide
level



Corrosives: Aqua Regia and Piranha

AVOID CONTACT WITH ORGANIC SOLVENTS

Aqua regia:

- 1:3 nitric to hydrochloric acid
 - use proper ventilation!
- $$\text{HNO}_3 + 3 \text{HCl} \rightarrow \text{NOCl} + \text{Cl}_2 + 2 \text{H}_2\text{O}$$
- don't use is closed containers
 - dilute with water slowly and neutralize

Piranha:

- 1:3 30% hydrogen peroxide to sulfuric acid
- mix slowly!
- dilute with water and neutralize

Try other cleaning methods first.

Other Common Dangerous Chemicals

Reproductive toxins

- carbon disulfide
- toluene
- pyridine
- HF

Acute toxins

- diazomethane
- sodium azide
- OsO₄
- organomercury
- organotin